This article was downloaded by:

On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



#### Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: <a href="http://www.informaworld.com/smpp/title~content=t713618290">http://www.informaworld.com/smpp/title~content=t713618290</a>

# Dimethyldithioimidocarbonates-Mediated Heterocyclizations: Synthesis of Imidazolidines and Benzheterocycles as Potent Antitubercular Agents

Vinayak S. Hegde<sup>a</sup>; Gundurao D. Kolavi<sup>a</sup>; Ravi S. Lamani<sup>a</sup>; Imtiyaz Ahmed M. Khazi<sup>a</sup> Department of Chemistry, Karnatak University, Dharwad, India

To cite this Article Hegde, Vinayak S. , Kolavi, Gundurao D. , Lamani, Ravi S. and Khazi, Imtiyaz Ahmed M.(2007) 'Dimethyldithioimidocarbonates-Mediated Heterocyclizations: Synthesis of Imidazolidines and Benzheterocycles as Potent Antitubercular Agents', Phosphorus, Sulfur, and Silicon and the Related Elements, 182: 4, 911 - 920

To link to this Article: DOI: 10.1080/10426500601088721

URL: http://dx.doi.org/10.1080/10426500601088721

#### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phosphorus, Sulfur, and Silicon, 182:911-920, 2007

Copyright © Taylor & Francis Group, LLC ISSN: 1042-6507 print / 1563-5325 online

DOI: 10.1080/10426500601088721



#### Dimethyldithioimidocarbonates-Mediated Heterocyclizations: Synthesis of Imidazolidines and Benzheterocycles as Potent Antitubercular Agents

Vinayak S. Hegde Gundurao D. Kolavi Ravi S. Lamani Imtiyaz Ahmed M. Khazi

Department of Chemistry, Karnatak University, Dharwad, India

A new series of thiadiazolylimidazolidines, thiadiazolylbenzimidazoles, thiadiazolylbenzoxazoles, and thiadiazolylbenzothiazoles were synthesized by heterocyclization reactions of dimethyldithioimidocarbonate of properly substituted thiadiazoles with various binucleophiles. The structures of all the newly synthesized compounds were elucidated and they were screened for antitubercular activity against Mycobacterium tuberculosis  $H_{37}$ Rv using the BACTEC 460 radiometric system, where few compounds have shown more than 90% inhibition at MIC of <6.25  $\mu$ g/mL in the preliminary screening level. They were also screened for their antibacterial activity against Escherichia coli and Bacillus cirrhosis, and antifungal activity against Aspergillus niger and Penicillium worthmanni. Some of the compounds have shown promising in vitro antibacterial and antifungal activities.

**Keywords** Antitubercular activity; antimicrobial activity; Dimethyldithioimidocarbonates; thiadiazolylimidazolidines; thiadiazolyl-benzheterocycles

#### INTRODUCTION

Thiadiazole derivatives are useful compounds having pharmacological and biological activities<sup>1-6</sup> such as antimicrobial,<sup>1-3</sup> anti-inflammatory,<sup>4,5</sup> anticonvulsant,<sup>6</sup> antihypertensive<sup>7</sup> and anticancer<sup>8,9</sup> activities. Furthermore, benzfused heterocycles. Particularly benzimidazoles, benzoxazoles, and benzothiazoles, are of considerable chemical and biological interest.<sup>10–13</sup> A combination of such two biodynamic

Received July 23, 2006; accepted October 5, 2006.

We sincerely thank Tuberculosis Antimicrobial Acquisition and Coordinating Facility (National Institute of Allergy and Infectious Diseases (NIAID), Maryland National Institutes of Health (NIH), Bethesda, Maryland, Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF), Birmingham, Alabama) for anti-tuberculosis evaluations and Dr. Uday Muddapur for antibacterial and antifungal screening. We also thank Prof. C. S. Mahajanashetti for valuable suggestions and encouragement.

Address correspondence to Imtiyaz Ahmed M. Khazi, Department of Chemistry, Karnatak University, Dharwad, 580003, India. E-mail:drimkogchem@gmail.com

molecules is expected to increase the biological potency of the molecule in concern. In recent years much attention has been devoted to the synthesis of heterocycles as antitubercular and antimicrobial agents. The treatment of mycobacterial infections, especially tuberculosis, has become an important problem due to the emergence of monodrug and multidrug resistance. <sup>13</sup> It has been reported that the thiadiazole derivatives showed good activity against *Mycobacterium tuberculosis*. <sup>15–17</sup>

#### **RESULTS AND DISCUSSION**

The new key intermediate dimethyl 5-(benzyl/3-pyridyl)-1,3,4-thiadiazol-2-yldithioimidocarbonates (2a,b) were synthesized from corresponding 2-amino-5-(benzyl/3-pyridyl)thiadiazoles (1a,b)<sup>18,19</sup> by the reaction with carbon disulfide and methyl iodide in alkaline medium (Scheme 1). The products are characterized by the absence of  $v_{\rm N-H}$  in the IR spectra and by the presence of a singlet for six protons (at  $\delta$  2.6 for 2a and  $\delta$  2.74 for 2b) due to two equivalent –SCH<sub>3</sub> groups in <sup>1</sup>H NMR spectra. The rest of the protons resonated in the expected region. Furthermore, compound 2b was confirmed by a <sup>13</sup>C NMR spectrum, which exhibited a characteristic signal at  $\delta$  16.4 due to carbon of two magnetically equivalent SCH<sub>3</sub> groups.

The intermediates dimethyldithioimidocarbonates (2a,b) were treated with ethylenediamine in DMF to yield corresponding imidazolidine derivatives (3a,b). The products obtained possessed the strong

R=(a) benzyl, (b) 3-pyridyl.

**SCHEME 1** Reagents and conditions: (a) i. aq. NaOH (20 M), CS<sub>2</sub>,  $0^{\circ}$ C/DMF, ii. CH<sub>3</sub>I,  $0^{\circ}$ C; (b) ethylene diamine/DMF,  $120^{\circ}$ C, 6 h; (c) 2-phenelenediamine/DMF,  $120^{\circ}$ C, 6–8 h; (d) 2-aminophenol/DMF,  $120^{\circ}$ C, 8 h; (e) 2-aminothiophenol/DMF,  $120^{\circ}$ C, 6–8 h.

 $\upsilon_{N-H}$  around the 3330 cm $^{-1}$  band in IR spectra, and the  $^{1}H$  NMR spectra displayed a  $D_{2}O$  exchangeable broad singlet in the region  $\delta$  7.3 to 7.7 assigned for two equivalent NH protons; a singlet around  $\delta$  3.6 for 4 protons was observed indicating the fact that both hydrogens are located (Figure 1) on ring nitrogens, thus making the protons  $-CH_{2}CH_{2}-$  of imidazolidine equivalent. Furthermore the  $^{13}C$  NMR spectrum of imidazolidine 3b showed a characteristic signal at  $\delta$  37.5 for  $-CH_{2}-CH_{2}-$  confirming both carbons are equivalent in accordance with the  $^{1}H$  NMR report. Compound 3b is also confirmed by mass (ESI) spectrum.

Similarly, when intermediate dimethyldithioimidocarbonates (2a,b) reacted with binucleophiles viz. o-phenylenediamine, 2aminophenol and 2-amino thiophenol afforded corresponding benzimidazoles (4a,b), benzoxazoles (5a,b), and benzothiazoles (6a,b), respectively. From the <sup>1</sup>H NMR spectrum of benzimidazole (4b), it is clear that both NH protons are equivalent. This is due to the reason as described in the case of imidazolidine derivatives (3a,b). Furthermore, this is supported by the fact that, in the <sup>1</sup>H NMR spectra of benzimidazole derivatives (4a,b), C<sub>5</sub>-H, and C<sub>6</sub>-H of benzimidazole ring was observed as a doublet at  $\delta$  7.2 and C<sub>4</sub>, C<sub>7</sub>-H was observed as a doublet at  $\delta$  7.4. The <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra of all benzheterocycles are in total agreement with the assigned structures. The mass spectrum of benzoxazole derivative **5a** showed a molecular ion peak at 308.1 (100%), which corresponds to its molecular weight (308.37). The new heterocyclic systems (3-6) were successfully constructed by the 4+1 approach, and the evolution of methanethiol (MeSH) was observed during the course of heterocyclization reactions.

#### **BIOLOGICAL ACTIVITY**

The in vitro antimycobacterial activity was assayed by Tuberculosis Antimicrobial Acquisition and Coordinating Facility (Birmingham, Alabama, USA), an antitubercular drug discovery program. Primary screening was conducted at 6.25  $\mu$ g/mL against M. tuberculosis  $H_{37}Rv$  (ATCC 27294) in BACTEC 12B<sup>20,21</sup> medium using a broth microdilution assay,  $^{22}$ the Microplate Alamar Blue Assay (MABA). Compounds exhibiting fluorescence were tested in the BACTEC 460 radiometric

**FIGURE 1** The delocalization of hydrogens over nitrogen atoms.

Compound		MIC	%	
no.	Assay	in $\mu$ g/mL	Inhibition	Activity
2a	Alamar	>6.25	88	_
2b	Alamar	>6.25	73	_
3a	Alamar	>6.25	77	_
3b	Alamar	>6.25	37	_
4a	Alamar	>6.25	77	_
4b	Alamar	> 6.25	74	_
5a	Alamar	>6.25	93	+
5b	Alamar	>6.25	95	+
6a	Alamar	> 6.25	53	_
6b	Alamar	>6.25	61	_

TABLE I The *In Vitro* Antitubercular Activity of Compounds Against *M. tuberculosis* H<sub>37</sub>Rv

system. Compounds affecting <90% inhibition in the primary screen were not further evaluated. Compounds demonstrating at least 90% inhibition in preliminary screening were tested at lower concentrations by serial dilution against M. tuberculosis  $H_{37}Rv$  to determine the MIC using MABA. Compounds 5a and 5b exhibited excellent inhibitory activity of 93% and 95%, respectively, which were selected for further screening, where it exhibited promising inhibitory activity of 58% and 39%, respectively.

Preliminary antituberculosis activity data (Table I) reveals that intermediate dimethyl-5-alkyl/aryl-1,3,4-thiadiazol-2-yldithioimidocarbonates (**2a,b**) are very active, and out of all heterocyclized derivatives, only benzoxazole derivatives (**5a,b**) have shown very good activity against M.  $tuberculosis~H_{37}Rv$  at  $<6.25~\mu g/mL$ . Other compounds have shown moderate activity.

Biological significance of all the derivatives was also established by screening them against two bacterial strains  $Bacillus\ cirrohsis$  and  $Escherichia\ coli$  and two fungal species  $Aspergillus\ niger$  and  $Penicillium\ wortmannii$  by the cup plate method 19 at the concentration 25–100  $\mu$ g/mL using DMF as a solvent. Norfloxacin and Greseofulvin, respectively, were used as standards. In general, antimicrobial activity data (Table II) revealed that among all the newly synthesized compounds, pyridine-substituted (at C-5 position of thiadiazole ring) compounds have shown very good antibacterial and antifungal activity at 100  $\mu$ g/mL concentration and moderate activity at a lower concentration of 25  $\mu$ g/mL. In particular, benzimidazole derivatives (4a,b) have shown good antibacterial activity at a higher concentration.

TABLE II The In Vitro Antibacterial Activity and Antifungal Activity

	Anti	ibacterial	activity (S	Antibacterial activity (Standard: Norfloxacin	Norfloxaci	(u	Ant	ifungal ac	tivity (Sta	Antifungal activity (Standard: Greseofulvin)	eseofulvir	(1
	Escl	Escherichia col	ıli	Bacil	Bacillus cirrhosis	sis	Penicilli	Penicillium wortmannii	annii	Aspei	Aspergillus niger	şer
Compound	100	20	25	100	20	25	100	20	25	100	20	25
no.	Relative	Relative Inhibition (%)	(%) u	Relative	Relative Inhibition (%)	(%) u	Relativ	Relative Inhibition (%)	(%) u	Relative	Relative Inhibition (%	(%) u
2a	78	63	20	72	48	Ι	89	I	Ι	57	-	I
2b	85	89	22	92	54	42	85	65	52	78	64	28
3a	65	I	I	85	28	I	48	I	I	22	I	I
3b	98	89	62	88	89	99	92	62	55	75	25	I
4a	55	I	I	79	62	53	72	58	I	54	I	I
4b	75	64	58	84	89	99	90	72	65	74	54	48
5a	71	I	I	48	I	I	74	52	I	48	I	I
5b	85	72	29	72	52	I	72	99	42	78	64	99
6a	65	I	I	72	I	I	70	55	I	62	I	I
<b>q</b> 9	72	64	48	82	28	45	75	89	55	72	54	I

In conclusion the preliminary *in vitro* antituberculosis, antibacterial, and antifungal screening results of novel imidazolidine and benzheterocycle derivatives of 1,3,4-thiadiazole reported in the present article reveal that many of the compounds from the series have emerged as potent antitubercular, antibacterial, and antifungal agents endowed with moderate to good activity. Possible improvements in the activity can be further achieved by slight modifications in the new heterocyclic systems reported in this article. Our findings will have an impact on chemists and pharmacists for further investigations in this field in search of potent antitubercular and antimicrobial agents.

#### **EXPERIMENTAL**

Melting points were determined using an electric melting point apparatus (Shital scientific industries, Mumbai) and are uncorrected. IR spectra (KBr) were run on a Nicolet impact 410 FT-IR spectrometer ( $\nu_{\rm max}$  in cm $^{-1}$ ).  $^{1}{\rm H}$  NMR and  $^{13}{\rm C}$  NMR spectra were recorded in CDCl $_{3}$ , DMSO- $d_{6}$ , and TFA with TMS as an internal standard (chemical shift in  $\delta$ , ppm and J values in Hz) on a Brucker 300 MHz FTNMR spectrometer. Mass spectra were recorded on Thermo-Finnigan-MAT, Bremen (Model MAT8200) spectrometer. Electrospray ionization mass spectrum (ESI-MS) was recorded on a Quattro LCZ (Walters-Micromass, Manchester), and elemental analyses were carried out on a Heraus CHN rapid analyzer. Purity of the compound was checked by TLC. Nomenclature was made using Chemsketch software. All the reagents were of laboratory reagent quality and were used after purification.

# Preparation of Dimethyl 5-(benzyl/3-pyridyl)-1,3,4-thiadiazol-2-yldithio imidocarbonates (2a,b): General Procedure

To a well-stirred solution of 2-amino-5-(benzyl/3-pyridyl)-1,3,4-thiadiazole  $\bf 2$  (10 mmol) in dimethylformamide (8 mL) were added sodium hydroxide solution (20 M, 4 mL), carbon disulfide (1.52 g, 20 mmol), and methyl iodide (3.384 g, 24 mmol) in sequence at an interval of 30 min at 0–5°C. Stirring was continued for 4 h, during which the temperature was slowly raised to r.t. The mixture was then poured into cold water, and the resulting sticky solid was extracted with chloroform, washed with water and dried over anhydrous sodium sulphate. The solvent was removed under diminished pressure, and the residual solid was recrystallized from a chloroform-hexane mixture to yield the corresponding dimethyldithioimidocarbonate ( $\bf 2a,b$ ) in moderate yields as crystalline solids.

### Dimethyl 5-benzyl-1,3,4-thiadiazol-2-yldithioimidocarbonate (2a)

Pale yellow needles (chloroform + hexane), yield 61%, m.p. 88–90°C; IR (KBr)  $\nu$ cm<sup>-1</sup>: 3052, 2977, 2925, 1608, 1574; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 2.59 (s, 6H, SCH<sub>3</sub>), 4.35 (s, 2H, CH<sub>2</sub>), 7.25–7.36 (m, 5H, phenyl). Anal. calcd. for C<sub>12</sub>H<sub>13</sub>N<sub>3</sub>S<sub>3</sub>: C, 48.78; H, 4.44; N, 14.22. Found: C, 48.93; H, 4.50; N, 14.08%.

### Dimethyl 5-pyridin-3-yl-1,3,4-thiadiazol-2-yldithioimidocarbonate (2b)

Bright yellow granules, m.p. 130–132°C; IR (KBr)  $\upsilon$ cm<sup>-1</sup>: 3038, 2978, 2919, 1583, 1568; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 2.74 (s, 6H, SCH<sub>3</sub>), 7.47 (dd,  $J_{\text{H5H4}} = 4.8$  Hz,  $J_{\text{H5H6}} = 3.1$  Hz, 1H, C<sub>5</sub>-H, pyridine), 8.29 (d, J = 4.8 Hz, 1H, C<sub>6</sub>-H, pyridine), 8.68 (d, J = 4.4 Hz, 1H, C<sub>4</sub>-H, pyridine), 9.08 (s, 1H, C<sub>2</sub>-H, pyridine); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 16.4, 124.2, 127.4, 134.6, 148.6, 151.7, 162.9, 169.0 and 177.1. Anal. calcd. for C<sub>10</sub>H<sub>10</sub>N<sub>4</sub>S<sub>3</sub>: C, 42.53; H, 3.57; N, 19.84. Found: C, 42.76; H, 3.61; N, 19.70%.

# Preparation of 5-(benzyl/3-pyridyl)-*N*-imidazolidin-2-ylidene-1,3,4-thiadiazol-2-amine (3a,b): General Procedure

A mixture of **2** (2 mmol) and ethylenediamine (0.14 g, 2.4 mmol) in dimethylformamide (10 mL) was heated to 120°C for 6 h. The solid that separated after cooling was collected by filtration, washed with aq. ethanol, dried, and recrystallized from ethanol.

#### 5-Benzyl-N-imidazolidin-2-ylidene-1,3,4-thiadiazol-2-amine (3a)

Colorless cubes (ethanol), yield 65%, m.p. 198–200°C; IR (KBr)  $\nu cm^{-1}$ : 3343, 2976, 1626, 1558;  $^1H$  NMR (300 MHz, CDCl $_3$ )  $\delta$ : 3.67 (s, 4H, imidazolidine), 4.23 (s, 2H, CH $_2$ ), 7.24–7.35 (m, 5H, phenyl), 7.72 (s, 2H, NH, D $_2$ O exchangeable);  $^{13}C$  NMR (75 MHz, CDCl $_3$ )  $\delta$ : 37.3, 42.7, 127.4, 129.10, 129.14, 137.9, 161.5, 161.6 and 176.4. Anal. calcd. for  $C_{12}H_{13}N_5S$ : C, 55.58; H, 5.05; N, 27.01. Found: C, 55.74; H, 5.17; N, 26.88%.

# N-Imidazolidin-2-ylidene-5-pyridin-3-yl-1,3,4-thiadiazol-2-amine (3b)

Colorless needles (ethanol), yield 61%, m.p. 254–256°C; IR (KBr)  $\nu$ cm<sup>-1</sup>: 3335, 3046, 2962, 1618, 1572, 1527; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.58 (s, 4H, imidazolidine), 7.35 (s, 2H, NH, D<sub>2</sub>O exchangeable), 7.47 (dd,  $J_{H5H4} = 4.8$  Hz,  $J_{H5H6} = 7.3$  Hz, 1H, C<sub>5</sub>-H, pyridine),

8.29 (d, J=7.8 Hz, 1H, C<sub>6</sub>-H, pyridine), 8.68 (d, J=4.4 Hz, 1H, C<sub>4</sub>-H, pyridine), 9.08 (s, 1H, C<sub>2</sub>-H, pyridine); MS (ESI): 247.07 (M+H), 269.05 (M+Na), 515.12 (2M+Na). Anal. calcd. for C<sub>10</sub>H<sub>10</sub>N<sub>6</sub>S: C, 48.77; H, 4.09; N, 34.12. Found: C, 48.94; H, 4.17; N, 33.96%.

### Preparation of *N*-{5-(benzyl/3-pyridyl)-1,3,4-thiadiazol-2-yl}-1*H*-benzimidazol-2-amine (4a,b): General Procedure

A mixture of 2 (2 mmol) and o-phenylenediamine (0.26 g, 2 mmol) was heated (120°C) in dimethylformamide (10 mL) for 8 h. The separated solid was collected by filtration, washed with aqueous alcohol, dried, and recrystallized from a suitable solvent.

### N-(5-Benzyl-1,3,4-thiadiazol-2-yl)-1H-benzimidazol-2-amine (4a)

Colorless needles (DMF), yield 72%, m.p. 264–268°C; IR (KBr)  $\nu$ cm<sup>-1</sup>: 3228, 3060, 2976, 1608, 1576, 1544;  $^1$ H NMR (300 MHz, DMSO- $d_6$ + TFA)  $\delta$ : 4.28 (s, 2H, CH<sub>2</sub>), 7.22–7.67 (m, 9H, benzimidazole). Anal. calcd. for C<sub>16</sub>H<sub>13</sub>N<sub>5</sub>S: C, 62.52; H, 4.26; N, 22.78. Found: C, 62.41; H, 4.31; N, 22.63%.

### N-(5-Pyridin-3-yl-1,3,4-thiadiazol-2-yl)-1H-benzimidazol-2-amine (4b)

Bright yellow solid (DMF), yield 75%, m.p. >300°C; IR (KBr)  $\nu$ cm<sup>-1</sup>: 3242, 3054, 1616, 1585, 1547; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ) δ: 7.15 (d, J=3.3 Hz, 2H, C<sub>5</sub>, C<sub>6</sub>-H, benzimidazole), 7.39 (d, J=3.3 Hz, 2H, C<sub>4</sub>, C<sub>7</sub>-H, benzimidazole), 7.55 (t, J=7.0 Hz, 1H, C<sub>5</sub>-H, pyridine), 8.23 (d, J=6.9 Hz, 1H, C<sub>6</sub>-H, pyridine), 8.66 (d, J=6.9 Hz, 1H, C<sub>4</sub>-H, pyridine), 9.04 (s, 1H, C<sub>2</sub>-H, pyridine), 12.06 (br s, 2H, NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ + TFA) δ: 110.0, 113.1, 113.8, 117.6, 121.4, 124.5, 127.4, 129.2, 130.4, 141.3, 142.7, 146.0, 147.3 and 155.1. Anal. calcd. for C<sub>14</sub>H<sub>10</sub>N<sub>6</sub>S: C, 57.13; H, 3.42; N, 28.55. Found: C, 57.25; H, 3.30; N, 28.38%.

# Preparation of *N*-{5-(benzyl/3-pyridyl)-1,3,4-thiadiazol-2-yl}-1,3-benzoxazol-2-amine (5a,b): General Procedure

A mixture of dimethyldithioimidocarbonate 2 (2 mmol) and 2-aminophenol (0.26 g, 2.4 mmol) in dimethylformamide (12 mL) was heated at  $120^{\circ}$ C for 8 h. The separated solid was collected by filtration, washed with aqueous alcohol, dried and recrystallized from a methanol-dimethylformamide mixture.

#### N-(5-Benzyl-1,3,4-thiadiazol-2-yl)-1,3-benzoxazol-2-amine (5a)

Pale yellow prisms (methanol + DMF), yield 69%, m.p. 282–286°C; IR (KBr) vcm<sup>-1</sup>: 3164, 3036, 1595, 1578, 1545;  $^{1}$ H NMR (300 MHz, DMSO- $d_6$ )  $\delta$ : 4.27 (s, 2H, CH<sub>2</sub>), 7.12–7.49 (m, 9H, benzoxazole), 13.95 (s, 1H, NH, D<sub>2</sub>O exchangeable);  $^{13}$ C NMR (75 MHz, DMSO- $d_6$  + TFA): 36.3, 110.6, 113.8, 116.0, 117.6, 121.4, 123.7, 125.2, 129.5, 129.6, 137.0, 138.2, 148.0 and 163.4; MS m/z (%): 308.0 (100), 270.1 (2.9), 248.0 (8.6), 177.0 (19.7), 160.1 (47.7). Anal. calcd. for  $C_{16}H_{12}N_4OS$ : C, 62.32; H, 3.92; N, 18.17. Found: C, 62.58; H, 3.81; N, 18.05%.

### N-(5-Pyridin-3-yl-1,3,4-thiadiazol-2-yl)-1,3-benzoxazol-2-amine (5b)

Pale yellow granules (methanol + DMF), yield 70%, m.p. >300°C; IR (KBr)  $\upsilon$ cm<sup>-1</sup>: 3320, 3072, 1609, 1572; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ + TFA)  $\delta$ : 7.31–7.48 (m, 4H, benzoxazole), 7.56 (t, J = 7.4 Hz, 1H, C<sub>5</sub>-H, pyridine), 8.25 (d, J = 7.2 Hz, 1H, C<sub>6</sub>-H, pyridine), 8.66 (d, J = 7.0 Hz, 1H, C<sub>4</sub>-H, pyridine), 9.04 (s, 1H, C<sub>2</sub>-H, pyridine). Anal. calcd. for C<sub>14</sub>H<sub>9</sub>N<sub>5</sub>OS: C, 56.94; H, 3.07; N, 23.71. Found: C, 57.18; H, 3.12; N, 23.56%.

#### Preparation of N-{5-(benzyl/3-pyridyl)-1,3,4-thiadiazol-2-yl}-1,3-benzothiazol-2-amine (6a,b): General Procedure

A mixture of dimethyldithioimidocarbonate 2 (2 mmol) and 2-aminothiophenol (0.3 g, 2.4 mmol) in dimethylformamide (12 mL) was heated at  $120^{\circ}$ C for 6–8 h. The solid that separated was collected by filtration, washed with aqueous ethanol, dried, and recrystallized from a suitable solvent.

# N-(5-Benzyl-1,3,4-thiadiazol-2-yl)-1,3-benzothiazol-2-amine (6a)

Light green granules (DMF), yield 64%, m.p. 264–266°C; IR (KBr)  $\nu$ cm<sup>-1</sup>: 3331, 3052, 1620, 1584, 1546; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$  + TFA)  $\delta$ : 4.32 (s, 2H, CH<sub>2</sub>), 7.15–7.52 (m, 9H, benzothiazole). Anal. calcd. for C<sub>16</sub>H<sub>12</sub>N<sub>4</sub>S<sub>2</sub>: C, 59.23; H, 3.73; N, 17.27. Found: C, 59.48; H, 3.80; N, 17.12%.

# N-(5-Pyridin-3-yl-1,3,4-thiadiazol-2-yl)-1,3-benzothiazol-2-amine (6b)

Light green granules (DMF), yield 67%, m.p. 288–292°C; IR (KBr)  $v \, \mathrm{cm}^{-1}$ : 3326, 3048, 1622, 1585, 1545; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$ : 7.26 (t, J=7.5 Hz, 1H, C<sub>6</sub>-H, benzothiazole), 7.42 (t, J=7.5 Hz, 1H, C<sub>5</sub>-H, benzothiazole), 7.54–7.91 (m, 3H, C<sub>4</sub>, C<sub>7</sub>-H, benzothiazole);

C<sub>5</sub>-H, pyridine), 8.31 (d, J=7.2 Hz, 1H, C<sub>4</sub>-H, pyridine), 8.70 (d, J=7.0 Hz, 1H, C<sub>6</sub>-H, pyridine), 9.10 (s, 1H, C<sub>2</sub>-H, pyridine), 13.23 (br s, 1H, NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$  + TFA) δ: 109.8, 113.6, 117.5, 121.3, 122.5, 123.9, 127.1, 128.0, 130.3, 130.7, 140.5, 142.8, 143.2 and 154.8. Anal. calcd. for C<sub>14</sub>H<sub>9</sub>N<sub>5</sub>S<sub>2</sub>: C; 54.00, H, 2.91; N, 22.49. Found: C; 54.26; H, 3.02; N, 22.34%.

#### REFERENCES

- [1] K. Desai and A. J. Baxi, Indian J. Pharm. Sci., 54, 183 (1992).
- [2] M. G. Mamolo, L. Vio, and E. Banfi, Farmaco., 51, 71 (1996).
- [3] A. K. Gadad, C. S. Mahajanshetti, S. Nimbalkar, and A. Raichurkar, Eur. J. Med. Chem., 35, 853 (2000).
- [4] M. D. Mullican, M. W. Wilson, D. T. Connor, C. R. Kostlan, D. J. Schrier, and R. D. Dyer, J. Med. Chem., 36, 1090 (1993).
- [5] L. Labanauskas, V. Kalcas, E. Udrenaite, P. Gaidelis, A. Brukstus, and A. Dauksas, Pharmazie, 56, 617 (2001).
- [6] C. B. Chapleo, P. L. Myers, A. C. Smith, M. R. Stillings, I. F. Tulloch, and D. S. Walter, J. Med. Chem., 31, 7 (1988).
- [7] S. Turner, M. Myers, B. Gadie, S. A. Hale, A. Horsley, A. J. Nelson, et al., J. Med. Chem., 31, 906 (1988).
- [8] M. Miyamoto, R. Koshiura, M. Mori, H. Yokoi, C. Mori, T. Husegawa et al., Chem. Pharm. Bull., 33, 5216 (1985).
- [9] J. Y. Chou, S.Y. Lai, S. L. Pan, G. M. Jow, J. W. Chern, and J. H. Guh, Biochem. Pharmacol., 66, 115 (2003).
- [10] P. N. Preston, Chem Rev., 74, 279 (1974), and references therein.
- [11] I. Yalcin, I. Oren, E. Sener, A. Akin, and N. Ucarturk, Eur. J. Med. Chem., 27, 401 (1992).
- [12] O. Temiz, I. Oren, E. Sener, I. Yalcin, and N. Ucarturk, Il Farmaco., 53, 337 (1998).
- [13] R. Paramashivappa, P. Phani Kumar, P. V. Subba Rao, and A. Srinivasa Rao, Bioorg. Med. Chem. Lett., 13, 657 (2003).
- [14] A. Rattan, A. Kalia, and N. Ahmad, Emerg. Infect. Dis., 4, 195 (1998).
- [15] E. E. Oruc, S. Rollas, F. Kandemirli, N. Shvets, and A. S. Dimoglo, J. Med. Chem., 47, 6760 (2004).
- [16] A. K. Gadad, N. N. Noolvi, and V. K. Rajshekhar, Bioorg. Med. Chem., 12, 5651 (2004).
- [17] G. Kolavi, V. Hegde, I. Khazi, and P. Gadad, Bioorg. Med. Chem., 14, 3069 (2006).
- [18] P. Mishra, A. K. Shakya, and G. K. Patnaik, J. Ind. Chem. Soc., 67, 520 (1990).
- [19] J. Kosla, Arch. Pharm., 287, 12 (1954); Chem. Abstr., 50, 969h (1956).
- [20] L. Collins and S. G. Franzblau, Antimicrob. Agents Chemother., 41, 1004 (1997).
- [21] S. G. Franzblau, R. S. Witzig, J. C. McLaughlin, P. Torres, G. Madico, A. Hernandez, et al., J. Clin. Microb., 36, 362 (1998).
- [22] W. J. Suling, L. E. Seitz, V. Pathak, L. Westbrook, E. W. Barrow, S. Zywno-van-ginkel, et al., Antimicrob. Agents Chemother., 44, 2784 (2000).
- [23] E. P. Casman, Am. J. Clin. Path., 17, 281 (1947).
- [24] A. I. Barry, The Antimicrobial Susceptibility Test—Principles and Practices (English Language Book Society, London, 4th ed., pp. 80–93, 1975).